

Optimization of Polysaccharide Ultrasonic Extraction Conditions Using Purple Sweet Potato Tubers Based on Free Radical Scavenging and Glycosylation Inhibitory Bioactivities

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ABSTRACT

Background: The purple sweet potato, *Ipomoea batatas*, belongs to the family *Convolvulaceae*. It is one of the most widely consumed tubers in Asia and is found in many dishes. Many people with diabetes eat purple sweet potato tubers to help reduce blood glucose in China. **Objective:** To predict the ultrasonic conditions for getting the optimal *in vitro* antioxidant and antiglycated activity of ultrasonic extracted polysaccharides from purple sweet potato (*I. batatas*) tubers, the artificial neural network (ANN) regression models was used in this study. **Materials and Methods:** The antioxidant activity of polysaccharides was quantified by evaluating the hydroxyl radical scavenging effect after ultrasonic extraction, and the data were used in conjunction with optimized extraction conditions to train the predictive ANN models. **Results:** The following conditions were predicted to yield optimal hydroxyl scavenging activity: 200 W, 22°C, and 40 min. In contrast, conditions of 230 W, 22°C, and 50 min yielded the greatest inhibitory effect on albumin nonenzymatic glycosylation. The accuracy and predictive ability of the models ranged from good to excellent, as indicated by R^2 values ranging from 0.953 to 0.998. **Conclusion:** The results of the present study showed that ANN predictive models are useful in ultrasonic processing, which can rapidly and accurately predict the optimum extraction conditions for polysaccharides based on their antioxidant and antiglycated activities. In addition, the results of the present study suggest that the consumption of sweet potatoes may help reduce free radicals in the body and prevent or treat diabetes.

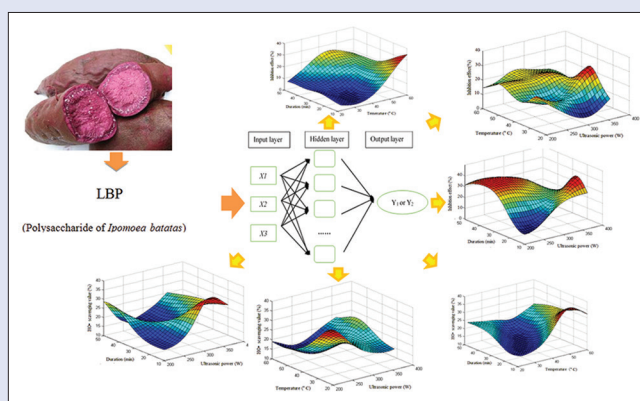
Key words: Antioxidant, *Ipomoea batatas* (L.) Lam, nonenzymatic glycosylation, polysaccharide, purple sweet potato

SUMMARY

- Ultrasonic extraction conditions were simulated and optimized using artificial neural network
- Bioactivities showed nonlinear relationship with ultrasonic conditions
- The optimal extraction conditions were 200 W, 22°C, and 40 min for the

highest antioxidant capacity

- The optimal extraction conditions were 230 W, 22°C, and 50 min for the highest antiglycated effect.



Abbreviations used: IBP: Polysaccharide of *Ipomoea batatas*; RSM: response surface methodology; ANN: Artificial neural network; BSA: Bovine serum albumin.

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INTRODUCTION

The purple sweet potato, *Ipomoea batatas*, known as “zishu” in mainland China, belongs to the family *Convolvulaceae*. It is one of the most widely consumed tubers in Asia and is commonly included in various dishes. The international project called Flora of China^[1] describes that “zishu” is a carbohydrate-rich food resource. The tubers of *I. batatas* Satsuma are purple and contain high levels of polysaccharides, anthocyanins, carotenoids, and polyphenols, in addition to dietary fiber, vitamins, and protein. Many of these compounds are regarded as antioxidants that eradicate free radicals.^[2-4] In China, eating purple sweet potato tubers could help reduce blood glucose levels in diabetes.

Previous studies have shown that components in the aqueous extracts of fruits and plants, including polysaccharides, have antioxidant, antiglycated, and antitumor activity.^[5-8] Recent epidemiological studies have associated the antioxidant ingredients in fruits, vegetables,

and plants with a reduced risk for life-threatening diseases such as diabetes, cancer, and stroke.^[9-12] The reasonable explanation is that most oxygen-based molecules from the atmosphere undergo oxidation to produce energy through the respiratory chain in the mitochondria; however, they often lose electrons and produce free radicals during

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metabolism, and make damage to cell and tissue; the levels of reactive oxygen species can significantly increase due to environmental stress.^[13,14] The hydroxyl molecule ($\bullet\text{OH}$) is one of the free radicals which can react with important biomolecules, resulting in cellular damage or death. Boligon *et al.*^[15] reported that the removal of $\bullet\text{OH}$ protects the brain and prevent diseases.

Central to our present understanding of diabetic hyperglycemia is the idea that high sugar levels lead to the covalent crosslinking of glucose, glycosylated protein, and lysine. Such crosslinks occur on the surface of proteins, influencing its normal function.^[16] Nonenzymatic glycosylation is the main biochemical reaction that generates these crosslinks. Studies by Brownlee *et al.*,^[17,18] inhibition of nonenzymatic glycosylation was shown to reduce the occurrence of protein glycosylation. Therefore, polysaccharides that inhibit glycosylation may have clinical applications that may potentially benefit patients with diabetic hyperglycemia.

Ultrasonication is commonly used technique for the extraction of natural products. With its unique advantages of mild extraction temperature, high extraction efficiency, this method can be applied to the extraction of compounds from a variety of animal tissues and plants. For these reasons, ultrasonication methods have to be used in extracting polysaccharides from sweet potato tubers. The main variable parameters in ultrasonic extraction are ultrasonication power, temperature, and duration; most of these parameters are associated with nonlinear, irregular effects,^[19] thus making it difficult to establish optimal conditions of extraction.

For solving this difficulty, we selected response surface methodology (RSM). The superior performance of RSM is reflected in the exploration of nonlinear or less systematic data. By training with several actual values obtained from varied conditions of ultrasonic extraction, the structured initial artificial neural network (ANN) model modified by genetic algorithms, it will transfer into a high-fitting class function that can accurately predict the outcomes of extraction conditions.^[20] Based on the self-adaptive, self-organizing, and self-learning characteristics of the ANNs of Wasserman,^[21] this model can analyze data with nonlinear or discrete distributions such as interactions between ultrasonic extraction conditions and bioactivities, thus improving data analysis and optimization of polysaccharide extraction conditions. Therefore, predicting the optimal ultrasonic conditions by genetic algorithms adjusted ANN model could solve the difficult to establish optimal conditions of extraction.

One folklore activity involves the consumption of sweet potatoes to reduce blood sugar levels of individuals with diabetes, which in turn has prompted researchers to determine the antioxidant effects of sweet potato extracts.^[22] The present study aimed to optimize the extraction of polysaccharides from purple sweet potatoes and to analyze the *in vitro* activities of extracted polysaccharides. The findings of the present study may serve as reference for future studies on sweet potato polysaccharides in relation to pharmacological issues.

MATERIALS AND METHODS

Materials

Raw tubers of the purple sweet potato (*I. batatas*) were purchased from a market at the Higher Education Mega Center (Guangzhou, China). Vitamin C, sodium azide (NaN_3), glucose standard, bovine serum albumin (BSA), and amino guanidine were purchased from Sigma-Aldrich (St. Louis, MO, USA). Other reagents were of analytical grade and were purchased from Guangzhou Chemical Reagent Factory (Guangzhou, China).

Preparation of purple sweet potato flour

Purple sweet potato tubers were cut into strips and completely dried in a 60°C air dryer. Dried samples were then pounded into a powder for use in subsequent experiments.

Extraction of polysaccharides from purple sweet potato tubers

Polysaccharides were extracted from 10 g of purple sweet potato powder dissolved distilled water (150 mL) in an ultrasonic cleaner (SB-4200DTD, Cosette Huai Instrument Co., Shanghai, China, 40 kHz) using different ultrasonication powers (200–360 W), temperatures (20–60°C), and duration (10–50 min), as described in Table 1. Each extract was filtered through a filter paper using a vacuum pump. Then, the filtrate was evaporated under vacuum in a rotary evaporator at 100 rpm and 65°C up to a volume of about 30 mL. Removal of protein from the filtrate was then conducted using the method of Sevag.^[19] The Sevag reagent (80 mL of CHCl_3 and 20 mL of butanol) was added to the concentrated extract, which was then shaken for 20 min at room temperature. After centrifugation at 3000 rpm for 20 min, the supernatant was collected. The Sevag reagent was then removed through dialysis, and 95% alcohol was added to yield a solution with an alcohol strength of up to 75%, as determined by a YH109 instrument (YanHe Instrument Co., Henan, China). The solution was then incubated overnight at 4°C to allow precipitation of polysaccharides. The resulting slurries were centrifuged at 3,000 rpm for 20 min, and the precipitates were collected. The precipitates were then lyophilized for 72 h at -47°C . Each lyophilized sample was weighed and placed in an individual Ziploc plastic bag until analysis. Polysaccharide content of the lyophilized purple sweet potato samples was determined by using the phenol-sulfuric acid method,^[4,23] using glucose as standard, and the results were expressed as glucose equivalents. The antioxidant capacity and inhibitory effects of the extracts on albumin nonenzymatic glycosylation *in vitro* were determined by conducting hydroxyl radical scavenging assays^[24] and the method of Brownlee *et al.*^[17,18] The optimal conditions for antioxidant and antiglycated activities were selected as described in the RSM illustration section.

Determination of three levels of three independent variables (X1, X2, and X3)

Three levels for each independent variable were selected using the procedure described. Three levels for each independent variable were determined by conducting a series of extractions in 150 mL of distilled water using different durations (10 min–50 min; variable X1), ultrasonication power (200 W–360 W; variable X2), and temperature (20°C–60°C; variable X3), as shown in Table 1. The full factorial design for the three variables (referred to as the 2^3 design) that describes the interactions in a total of 27 experimental groups is shown in Table 2.

Characterization of the extracted polysaccharides from sweet potato tubers.

Table 1: The three selected levels of three independent variables in this experiment

Variable	Symbol	Variable levels		
Extract time (min)	X1	10	30	50
Ultrasonic power (W)	X2	200	280	360
Extract temperature (°C)	X3	20	40	60

Table 2: Experiments for polysaccharide extraction in a 3×3×3 full factorial design

Experiment (n) ^a	Ultrasonic power (W)	Extract temperature (°C)	Extract time (min)
1	360	60	10
2	360	60	30
3	360	60	50
4	280	60	10
5	280	60	30
6	280	60	50
7	200	60	10
8	200	60	30
9	200	60	50
10	200	40	10
11	200	40	30
12	200	40	50
13	280	40	10
14	280	40	30
15	280	40	50
16	360	40	10
17	360	40	30
18	360	40	50
19	200	20	10
20	200	20	30
21	200	20	50
22	280	20	10
23	280	20	30
24	280	20	50
25	360	20	10
26	360	20	30
27	360	20	50

^aExperiments were conducted in a random order

Content analysis of each polysaccharide

The following glucose standards were used in the analysis: 0.02, 0.04, 0.05, 0.06, 0.07, 0.08, 0.10, and 0.12 mg/mL. Next, 1 mL of each standard was transferred into a tube. For conditions using ice water, 1 mL of a 5% phenol solution was slowly added, and the sample was mixed. Next, 5 mL of concentrated sulfuric acid was rapidly added, and the solution was shaken and incubated for 20 min at room temperature. For the blank control, 1 mL of distilled water was added using the same procedure. The absorbance was measured at a wavelength of 490 nm using a spectrophotometer. Glucose content was determined using Equation 1, which was derived from the standard curve of glucose:

$$Abs = 10.0425c + 0.0018 \quad R^2 = 0.9989 \quad (1)$$

Approximately, 10 mg of the lyophilized polysaccharide powder was mixed with water to a total volume of 100 mL to yield a final concentration of 0.1 mg/mL. Next, 1.0 mL of the solution was transferred into a tube, and the same procedure was performed as earlier described. Glucose concentration was calculated by using the absorbance of the solution and Equation 1. Percent glucose content was calculated using Equation 2:

$$\text{Glucose content (\%)} = \frac{c}{0.1 \text{ mg/mL}} \times 100\% \quad (2)$$

where c is the concentration of glucose calculated by the glucose standard curve equation above.

Hydroxyl radical scavenging activity assay

The hydroxyl radical scavenging activity of each extract was measured by using a modified Cumbe and Smironoff method and the Fenton reaction.^[25] The reaction mixture (10 mL) contained 2 mL of FeSO₄ (9 mM), 2 mL of salicylic acid (9 mM), and 2 mL of the sample (0.5 mg/mL). For positive control, 2 mL of 0.5 mg/mL ascorbic acid was used; and for blank solvent control, 2 mL of distilled water was

employed. Approximately, 2 mL of H₂O₂ (8 mM) was added to activate the reaction, followed by incubation of the solution at 37°C for 0.5 h. The absorbance of each solution was measured at a wavelength of 510 nm using an ultraviolet spectrophotometer. The hydroxyl radical scavenging capacity was calculated using Equation 3:

$$\text{Scavenging effect (\%)} = \left(1 - \frac{A_{\text{sample}}}{A_{\text{control}}}\right) \times 100\% \quad (3)$$

where A_{control} is the absorbance of the control, and A_{sample} is the absorbance of the samples.

Analysis of the inhibitory effects of sweet potato tuber extracts on albumin nonenzymatic glycosylation *in vitro*.

The inhibitory effects of the sweet potato tuber extracts were examined by using a modified version of the method of Brownlee *et al.*^[17,18,26] A glucose mixture containing 5% (w/v) BSA, 1 M glucose, and 0.1% (w/v) NaN₃ (pH 7.4) in phosphate-buffered saline (PBS) was used. Each lyophilized polysaccharide extract was dissolved to yield a 0.5 mg/mL solution. (i) The same amount of distilled water was added to the blank control (o), or the same concentration of amino guanidine was used as positive control.

All of the solutions were filtered through a 0.22 μm Millipore filter before transferring into Eppendorf tubes. Each sample was incubated in the dark at 37°C for 4 weeks, and the fluorescence intensity was measured each week with a fluorospectrophotometer at an excitation wavelength (Ex) of 370 nm and an emission wavelength (Em) of 440 nm. The inhibitory effect was calculated by using Equation 4, and the value at the 4th week was used for RSM analysis.

$$\text{Inhibitory effect (\%)} = \left(1 - \frac{F_i}{F_o}\right) \times 100\% \quad (4)$$

where F_i is the fluorescence intensity of the lyophilized polysaccharide extract, and F_o is the fluorescence intensity of the blank control.

Design of the artificial neural network

Modeling of the artificial neural network

By using the back-propagation algorithm,^[27,28] the multilayer feed-forward and feedback neural network has been proven to be an excellent universal approximator of nonlinear functions.^[29] The architecture of this model represents a fully-connected structure, and each experimental group is considered as a neuron. In addition, each neuron is connected to all the neurons in the hidden layer.^[30] Figure 1 shows that in the present study, a feed-forward neural network trained with an error back-propagation algorithm was employed using MATLAB Neural Network Toolbox (Version 7.11, Mathworks, Natick, MA, USA) to model the hydroxyl radical scavenging value and glycosylation inhibition of independent variables used in the ultrasonic treatments. The input parameters chosen in the present study were ultrasonication power, extraction time, and temperature, and the output parameters were the hydroxyl radical scavenging value and the inhibitory effect on glycosylation value (Y_1 and Y_2 , respectively). The neurons in the input layer introduced the normalized input data to the hidden layer via weights. The neurons in the hidden layer summed up the weighted inputs to neurons, including bias, based on Equations 5–7:

$$\text{Net} = \sum_{i=1}^n x_i w_i + \theta \quad (5)$$

where w_i ($i = 1, n$) w_i ($i = 1, n$) is the connecting weight, θ is the bias, x_i is the input data, and net is the activation.

$$\text{Tansig}(\text{net}) = \frac{2}{1 + \exp(-2 \times \text{net})} - 1 \quad (6)$$

The weighted output was then activated using the hyperbolic tangent sigmoid transfer function, as shown in Equation 7.

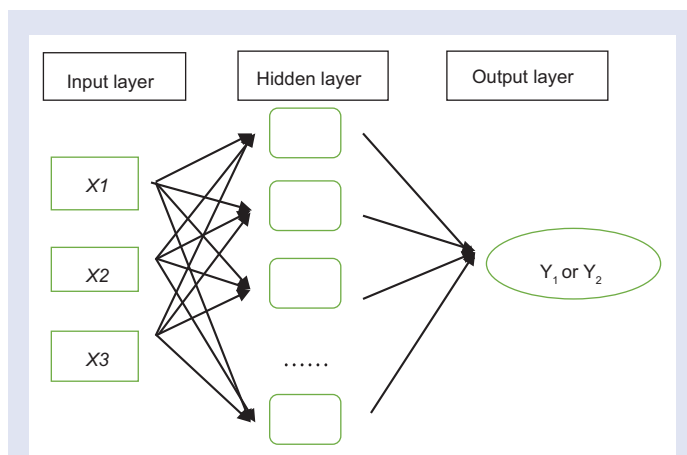


Figure 1: Diagram of the architecture of the artificial neural network model (not all the factors in the hidden layer are shown). Showing the internal connections among extraction time (X_1), ultrasonic power (X_2), and extraction temperature (X_3) with hydroxyl radical scavenging activity (Y_1) and nonenzymatic glycosylation inhibition (Y_2)

$$\text{Pureline } (x_i) = x_i \quad (7)$$

The output produced by the hidden layer became an input to output using this linear transfer function.

Supervised learning was used to train this network. The predicted output and desired output were compared to a virtual value and the errors were calculated. An error back propagation algorithm was used to adjust the weights to reduce the deviations.^[31] Using this decreasing gradient, the weights were proportionally changed to decrease the error gradient. The training iterations were terminated when the validation error reached the set minimum.

Optimization of the artificial neural network

Using the genetic algorithms of Goldberg and Holland,^[32] Yang *et al.*^[19] solved linear and nonlinear problems by exploring all aspects of the network and scanning promising areas through selection, multiplication, and mutation operations applied to individuals in the population.^[3] Genetic algorithms also reduce the number of times required for training iterations, thereby avoiding training to an extreme microregion. After the feed-forward ANN model was completely trained, genetic algorithms can identify the optimal ultrasonic extraction conditions that would yield the highest hydroxyl radical scavenging value or the strongest inhibitory effects on glycosylation.

Statistical analysis

All data were expressed as the mean of three replicates. Statistical calculations were performed using Microsoft Excel 2013 (Microsoft, Seattle, WA, USA) and MATLAB.

RESULTS AND DISCUSSION

Antioxidant activities, inhibition of glycosylation, and polysaccharide content of each sweet potato tuber extract.

Polysaccharides from purple sweet potato tubers contain a high number of hydroxyl groups that have the capacity of eradicate free radicals. According to Zhu *et al.*,^[33] hydroxyl radicals are produced through several different reactions. Using the products of the Fenton reaction to produce free radicals *in vitro*, we measured the hydroxyl radical scavenging activity of the extracted polysaccharides. Table 3 shows the hydroxyl radical scavenging activity of extracts prepared using different extraction

conditions. Ultrasonic treatment facilitates in the destruction of plant cells and influences the release of polysaccharides. Under the conditions tested in the present study, the scavenging rates of the extracts ranged from 11.22% to 30.89%. These findings suggest that the consumption of purple sweet potato tubers may reduce the amount of free radicals present in the body. Polysaccharides from purple sweet potatoes may also have some antitumor activities, similar to the polysaccharides isolated from sclerotia and mycelia of *Pleurotus tuber-regium*.^[7]

The Brownlee method is a conventional technique for the identification of AGE inhibitors that can block the production of early glycosylation products. Amino guanidine was used as a positive control in the inhibitory effects assay.^[24,34] Table 3 shows that all lyophilized sweet potato samples inhibited glycosylation when used at a concentration of 0.5 mg/mL. Differences in the inhibitory effects among groups may have resulted from variations in extraction conditions, which in turn might have influenced polysaccharide dissolution. Nonetheless, the results showed that polysaccharides from purple sweet potatoes inhibited nonenzymatic glycosylation and may be potentially used for the prevention and/or treatment of diabetes.^[35]

Polysaccharide content of the experimental groups was determined by using the phenol-sulfuric acid method, and each sample contained about 39.66%–78.62% deoxidized sugar. Further studies on the relationship between glucose content and antioxidant activity or glycosylation inhibitory activity are in progress. There may be other nonreducing sugars that affect the antioxidant and glycosylation inhibitory activities of these extracts.

All of these three assays yielded data that were nonlinear in nature and described its relationship with various ultrasonic extraction conditions. Thus, these assays, which used ANN through the RSM in MATLAB, provided significant improvement in this particular extraction methodology.

Optimization and prediction of ultrasonic extraction conditions by ANN according to hydroxyl radical scavenging activity and non-enzymatic glycosylation inhibition.

Table 3 and Figure 2 show the nonlinear relationship of ultrasonication temperature, time, power, and different combinations of these variables on hydroxyl radical scavenging activity. The optimum conditions for ultrasonic extraction were as follows: Ultrasonication power at 200 W, temperature at 22°C, and duration of 40 min. These conditions are predicted to result in extracts with 38.3% hydroxyl radical scavenging activity. Validation experiments showed an activity of 39.0% [verification set No. 28, Table 3].

Figure 2a-c shows that the highest free radical scavenging activity was obtained when the ultrasonication power and temperature were close to 200 W and 20°C, respectively. On the other hand, in terms of ultrasonication power and extraction time, the respective conditions of 320 W and 10 min showed the best results. In terms of extraction time and temperature, the optimal conditions were 50°C for 10–15 min. The more complicated interactions between two of the three factors were demonstrated in each three-dimensional graph. For example, when 40 min of ultrasonication at 50°C was performed, hydroxyl radical scavenging activity decreased with increasing ultrasonication power, and thereafter increased. On the other hand, lower scavenging activity was observed with increasing ultrasonication power for 40 min at 30°C. Similar relationships were observed between time and power, as well as between temperature and power.

Table 3 and Figure 3 show that the three factors interacted when the glycosylation inhibitory effect was utilized as output. The nonlinear nature of the relationships among the three conditions and the glycosylation inhibitory effect is shown in Figure 3a-c. Each of the three factors is

Table 3: The predicted and actual values of glycosylation inhibitory effects and hydroxyl radical scavenging activities of polysaccharides extracted from purple sweet potatoes prepared under different extraction conditions

Experiment (n) ^a	Ultrasonic power (W)	Extract temperature (°C)	Extract time (min)	Glycosylation inhibitory effect (%)		Hydroxyl radical scavenging effect (%)		Glucose content (%)
				Actual	Predicted	Actual	Predicted	
0	Comparison			68.60		95.06		
Training set								
1	200	20	10	13.50	12.98	23.87	23.91	74.33
2	200	60	10	13.30	12.92	30.89	30.89	47.76
3	200	60	30	13.00	11.70	20.19	20.11	42.44
4	200	60	50	15.10	13.94	30.02	30.03	57.77
5	200	40	10	6.20	6.41	33.30	33.42	50.93
6	200	20	30	17.40	16.36	14.15	14.02	76.11
7	200	20	50	20.20	23.09	18.36	14.29	76.65
8	200	40	30	7.69	9.75	13.93	14.09	51.17
9	280	60	30	17.60	17.85	22.79	22.90	53.87
10	280	40	10	11.80	12.80	26.13	25.94	39.66
11	280	40	50	20.90	20.71	18.47	18.51	72.27
12	280	20	30	22.00	22.77	19.11	18.82	66.90
13	280	20	10	27.10	24.98	16.42	16.37	64.11
14	280	60	10	8.20	7.26	22.89	24.10	40.26
15	280	20	50	14.30	14.19	26.57	26.58	73.30
16	280	60	50	24.40	25.89	21.38	21.42	59.96
17	360	20	10	19.10	20.20	25.70	24.54	64.09
18	360	60	50	17.50	17.70	20.63	20.78	51.10
19	360	60	30	14.60	14.17	30.24	30.23	44.27
20	360	40	10	15.70	15.79	11.23	11.24	47.95
21	360	40	30	18.80	18.36	26.35	26.29	65.44
22	360	40	50	20.90	20.14	16.09	16.06	73.04
23	360	20	50	14.00	13.96	27.32	27.29	63.18
Testing set								
24	360	20	30	18.10	17.95	26.35	26.36	66.44
25	360	60	10	23.10	23.84	20.95	20.95	56.97
26	280	40	30	16.30	16.30	27.54	27.35	58.81
27	200	40	50	3.90	3.58	23.54	23.48	56.91
Verification set								
28	200	22	40			39.00	38.30	78.62
29	320	26	10	29.12	31.90			63.41

^aExcept number 0, experiments number were conducted in a random order

presented in both the three-dimensional graph and its corresponding two-dimensional elevation chart. The optimal conditions for ultrasonic extraction, as computed through ANN, were as follows: Ultrasonication power at 230 W, temperature at 20°C, and duration of 50 min. The predicted hydroxyl radical scavenging activity was 34.5%. We predicted the powers of 200, 240, 280, 320, and 360, and the optimum condition of 320 W and 26°C for 10 min, to result in a 31.9% glycosylation inhibition, which was obtained from these five power values. Validation testing generated an inhibitory value of 29.1%, as shown in verification set No. 29 of Table 3.

Similarly, using the same ultrasonication power, a glycosylation inhibitory effect was observed, even when the variables of time and temperature were changed.

Overall, our data showed that at a certain ultrasonication power, the activity of antioxidants and the inhibition of glycosylation increased with higher temperatures (50°C–60°C) and shorter duration (10–25 min). Furthermore, these inhibitory effects increased with lower temperatures (20°C–40°C) and longer ultrasonication duration (30–50 min). We believe that similar results may be observed using other factors not included in the present study. These data are presented as two peak regions in the same figure.

Similar to the results of Li *et al.*,^[36] changes in the inhibitory effects arise from ultrasonic treatment, which could modify polysaccharide

solubility and yield during extraction. In addition, Wu *et al.*^[37] reported that the activities of polysaccharides largely depend on the number of hydroxyl groups and the structure of the actual molecule.^[20] In the present study, a higher ultrasonication power resulted in a greater reduction in molecular weight of the polysaccharide. However, at the same time, higher ultrasonication power increased molecular motion, thereby improving the solubility of the polysaccharide. In terms of extraction time, longer durations may result in more soluble polysaccharide extracts, but also increase the shearing effect of the ultrasonic waves. Higher temperatures also promote polysaccharide solubility, but again may accelerate the shearing effects of the ultrasonic waves. Therefore, because of the complicated relationships among ultrasonic extraction factors, RSM has facilitated in the optimization of extraction conditions for polysaccharides from sweet potato tubers.

Evaluation of model predictability

In the present study, RSM was employed to identify the optimal ultrasonication power, temperature, and time for achieving maximal hydroxyl radical scavenging activity and nonenzymatic glycosylation inhibition. The actual and predicted values are presented in Table 3.

The parameter R^2 is defined as the ratio of the explained variation to the total data and reflects the degree of fitness. The model well fits the

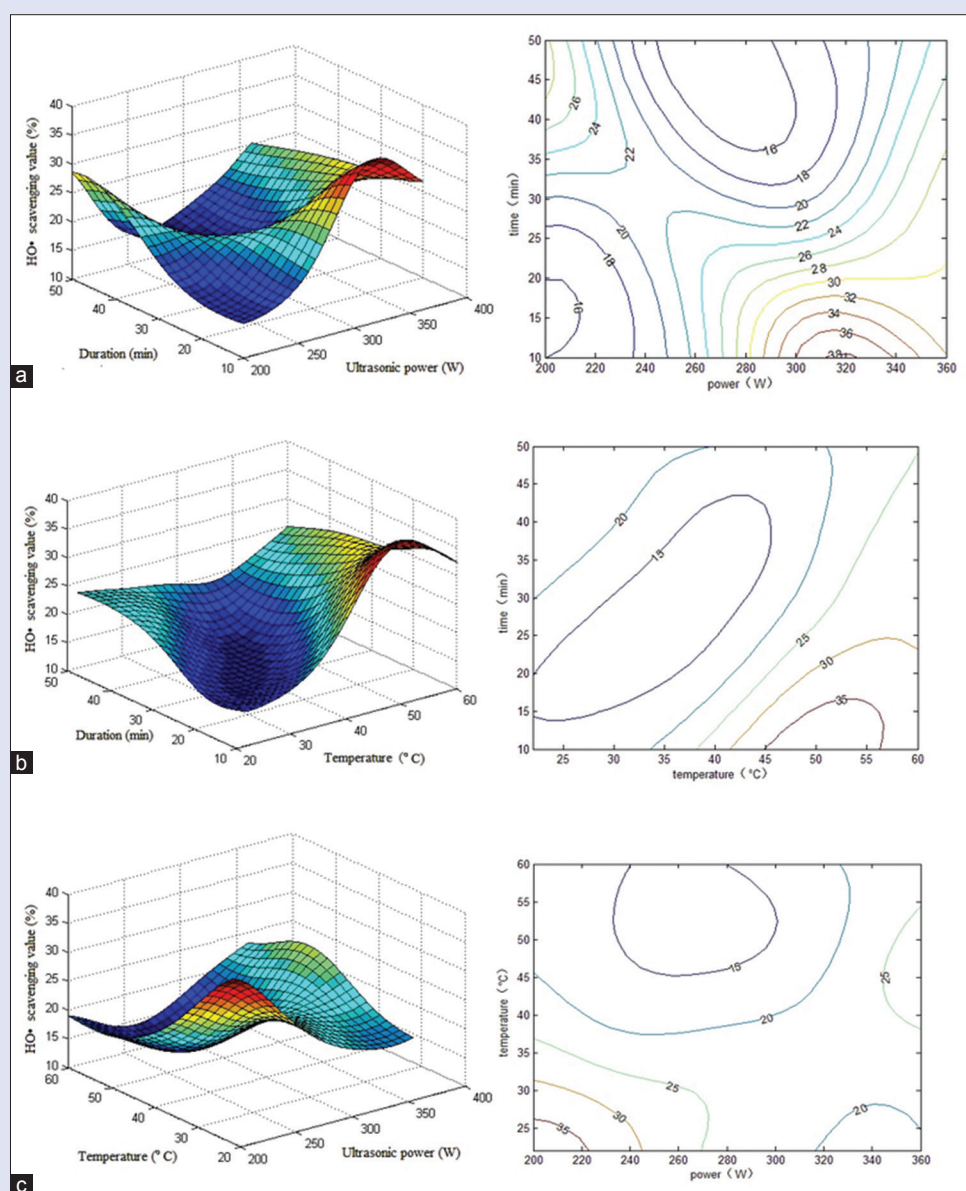


Figure 2: Response surface plot showing the effects of ultrasonication power and duration on hydroxyl radical scavenging activity. The corresponding two-dimensional elevation chart is shown on the left. Experimental and predicted data and conditions are shown in Table 3. (a) Ultrasonication temperature was kept constant at 22°C; (b) ultrasonication power was kept constant at 200 W; (c) ultrasonication temperature was kept constant at 40 min

actual values when R^2 approached 1. The performance of the ANN was statistically analyzed by using the R^2 values obtained using Equation 8:

$$R^2 = 1 - \frac{\sum_{i=1}^n (y_p - y_o)^2}{\sum_{i=1}^n (y_o - y_a)^2} \quad (8)$$

where n is the number of experimental groups, y_p is the predicted value by using the ANN model, y_o is the actual value, and y_a is the average of the actual values.

The ANN effects of the three extraction factors within specific ranges (ultrasonication power: 200–360 W, duration: 10–50 min, and temperature: 20°C–60°C) were examined. To test the generalization capacity of the network, 27 experimental groups were divided into two sets, and four groups were randomly selected [Table 3] for testing. The

rest of the sets was used for training. Data processing was performed using the least squares method in Microsoft Excel, and the results were analyzed as plots of the actual and predicted values of hydroxyl radical scavenging activities and R^2 values [Figure 4]. The R^2 values of the training and testing sets were 0.953 and 0.999, respectively. Using MATLAB, the trained network yielded an R^2 of 0.957. Figure 5 depicts plots of the actual and predicted values of nonenzymatic glycosylation inhibition, as well as the R^2 values. The R^2 values of the training and testing sets were 0.996 and 0.999, respectively. The trained network yielded an R^2 of 0.998 when using MATLAB.

Therefore, based on the MATLAB results, both ANNs predicted a value within the range of $\pm 0.8\%$ of the actual value. These findings were in good agreement with the predicted value of the neural network model and the actual value in the two model predictions, thereby confirming that this ANN is generally applicable.

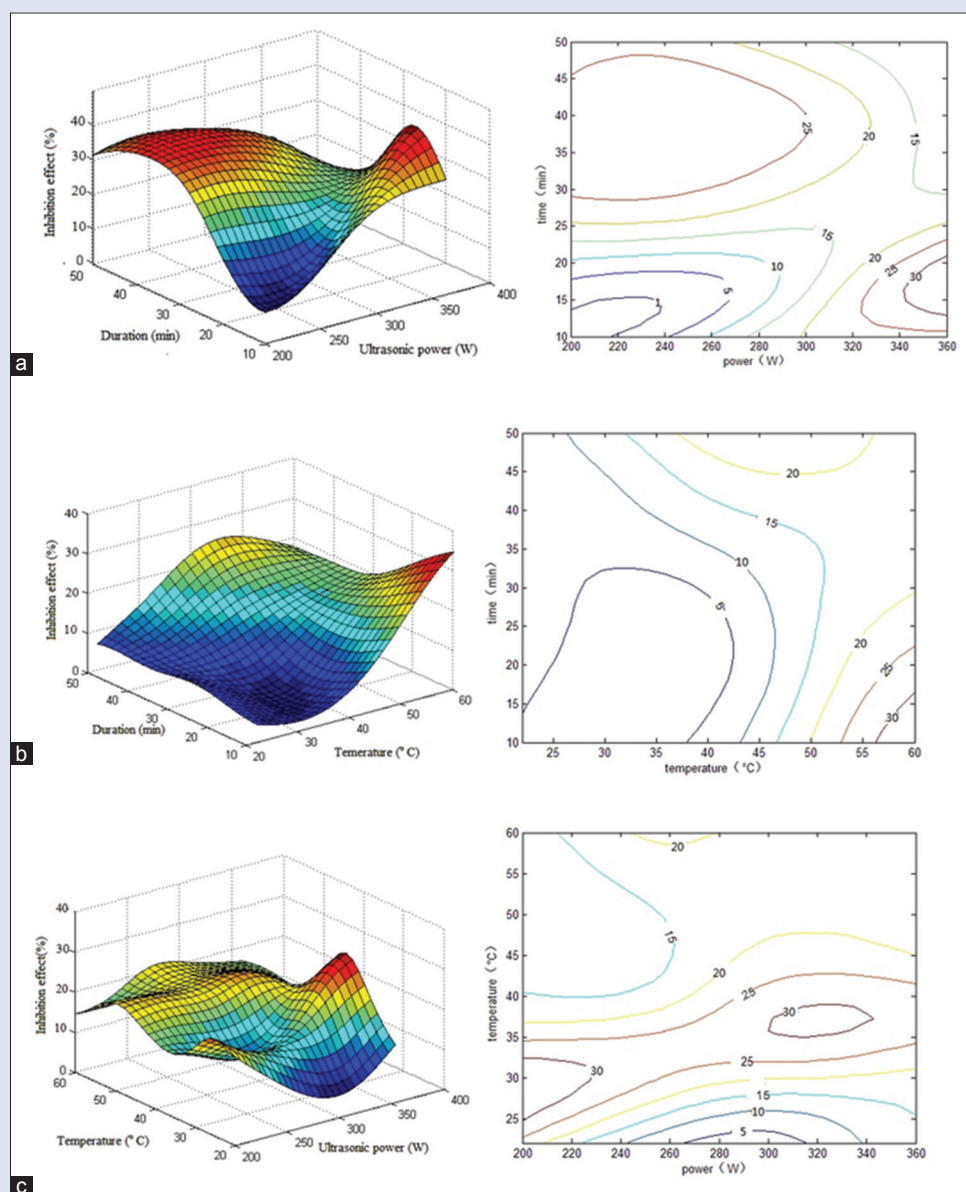


Figure 3: Response surface plot showing the effects of ultrasonication power and duration on the inhibition of nonenzymatic glycosylation. The corresponding two-dimensional elevation chart is shown on the left. Experimental and predicted data and conditions are shown in Table 3. (a) Ultrasonication temperature was kept constant at 22°C; (b) ultrasonication power was kept constant at 230 W; (c) temperature was kept constant at 50 min

CONCLUSIONS

Polysaccharides extracted from purple sweet potato tubers showed free radical scavenging activity and inhibited albumin nonenzymatic glycosylation. Using RSM, we developed an ANN model, which predicted high-fitting experimental data. Ultrasonication conditions of 200 W, 22°C, and 40 min generated the highest hydroxyl scavenging activity, whereas conditions of 230 W, 22°C, and 50 min yielded the greatest inhibitory effects on albumin nonenzymatic glycosylation. Multiple regression analysis indicated that ultrasonication power, temperature, and duration largely influenced the antioxidant and glycosylation inhibitory activities of these extracts. However, these activities did not show an absolute relationship with polysaccharide content. Perhaps, these were affected by other nonreducing sugars in the extracts. The results suggested that frequent consumption of purple sweet potatoes may facilitate in the reduction of blood sugar levels in patients with diabetes, which also further strengthens

the validity of this therapeutic folk diet. Further studies on the bioactivities of other polysaccharides in the purple sweet potato are in progress.

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Conflicts of interest

There are no conflicts of interest.

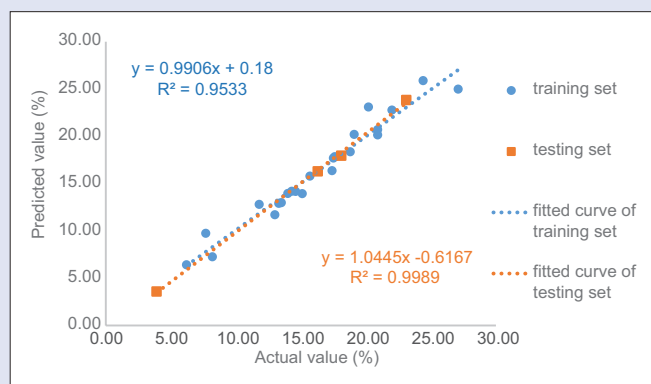


Figure 4: Correlation between the predicted values of the neural network and the actual values of the hydroxyl radical-scavenging activity. The training set and testing set were obtained from this chart. The fitting formula and its R^2 were included. Experimental and predicted data and conditions are shown in Table 3

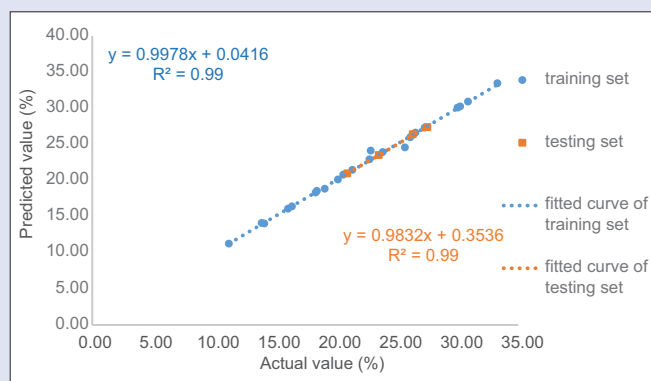


Figure 5: Correlation between the predicted values of the neural network and the actual values of the glycosylation inhibitory effect. The training set and testing set were obtained from this chart. The fitting formula and its R^2 were included. Experimental and predicted data and conditions are shown in Table 3

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