

A New Biphenyl Neolignan from Leaves of *Patrinia villosa* (Thunb.) Juss.

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ABSTRACT

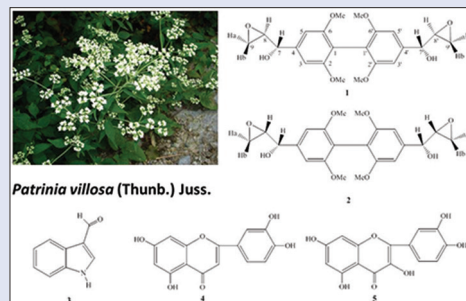
Results: One new stereoisomer of biphenylneolignan with four known compounds was isolated from the leaves of *Patrinia villosa* Juss.

Methods: The structure of the new compound was elucidated as 2,6,2',6'-tetramethoxy-4,4'-bis (1,2-trans-2,3-epoxy-1-hydroxypropyl) biphenyl (1) on the basis of spectroscopic analysis and comparison with literature data. The four known compounds were identified as 2,6,2',6'-tetramethoxy-4,4'-bis(1,2-cis-2,3-epoxy-1-hydroxypropyl)biphenyl (2), 1H-indole-3-carbaldehyde (3), luteolin (4) and quercetin(5) by comparison of their spectral data with the reported data, respectively. **Conclusions:** Compound 1 is a new biphenylneolignan, compound 2 and 3 were isolated for the first time from the plant.

Key words: Biphenyl neolignan, coupling constant, nuclear magnetic resonance, *Patrinia villosa* (Thunb.) Juss., structure

SUMMARY

- One new stereoisomer of biphenylneolignan named 2,6,2',6'-tetramethoxy-4,4'-bis (1,2-trans-2,3-epoxy-1-hydroxypropyl) biphenyl with four known compounds was isolated from the leaves of *Patrinia villosa* Juss.



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INTRODUCTION

Patrinia, which belongs to the family of Valerianaceae, is a widely distributed plant grown in East Asia and North America. The genus included more than 20 species, 10 of which growing in China.^[1] Usually, *Patrinia* species are used as leaves and vegetables in some areas of China, and research also revealed its leaves with pharmacological properties, especially the species of *Patrinia villosa* (Thunb.) Juss. The *P. villosa* (Thunb.) Juss., is an important ancient herbal medicine widely used for more than 2000 years from Shen Nong Ben Cao Jing, a famous ancient Chinese medicinal literary. It has been used in traditional Chinese medicine for an inflammation, wound healing, and abdominal pain. The previous research on the chemical constituents of *P. villosa* (Thunb.) Juss., have revealed that it contains several compound classes. Pentacyclic triterpenoids, iridoids, and flavonoids are the dominant bioactive constituents in the leaves of *P. villosa* Juss., which displayed potential ability of anti-tumor and anti-inflammatory.^[2-5] Other components, such as sterols, fatty acids, and essential oil^[2,6-8] were also confirmed in *P. villosa* (Thunb.) Juss.

As part of our search for new natural compounds with anti-tumor activity, we carried out phytochemical investigations on the leaves of *P. villosa* (Thunb.) Juss., collected in China. A new biphenyl neolignan and four known compounds [Figure 1] were isolated from the plant. The present study reports the isolation and structural elucidation of these components.

MATERIALS AND METHODS

General experimental procedures

Optical rotations were measured on a Perkin-Elmer digital polarimeter (USA). nuclear magnetic resonance (NMR) spectra were recorded

with tetramethylsilane as internal standard on a Bruker AVANCE 400 FT-NMR spectrometer. Electrospray ionization/mass spectrometry was measured on an Agilent 1100 LC/MSD Trap-SL spectrometer (USA). Column chromatography was performed on silica gel (200–300 mesh) (Marine Chemical Factory, Qingdao, China) and octa decylsilyl silicium (ODS) (ultimate XB-C18, 40–70 μm, Welch Materials, Inc., USA). High-performance liquid chromatography (HPLC) separation was performed with an Agilent 1100 chromatograph apparatus using an ODS column (ultimate XB-C18, 10 × 250, 5 μm, Welch Materials, Inc., USA) and detected with an ultraviolet detector.

Materials and chemicals

The *P. villosa* Juss., leaves were purchased from Hebei Qixin Traditional Chinese Medicine Pellets Co., Ltd., P.R. China, and identified by Prof. Kang Tingguo, College of Pharmacy, Liaoning University of Traditional Chinese Medicine. A voucher specimen (NO. PVJ20130821) has been deposited at the Pharmacognosy Laboratory, Harbin University of

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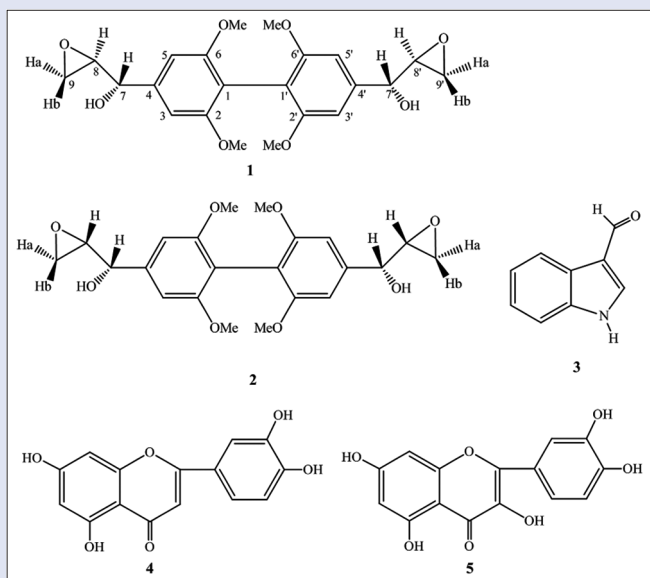


Figure 1: The structures of compounds 1-5

Commerce. Organic solvents (analytical grade or HPLC grade) for the experiment were purchased from Kermel Chemical Co., (Tianjin, China).

Extraction and isolation

The air-dried leaves of *P. villosa* Juss., (15 kg) were extracted three times with 70% EtOH under reflux. Evaporation of the solvent under reduced pressure gave a condensed extract (2.8 kg), which was suspended in H₂O and then partitioned successively with light petroleum, dichloromethane, and *n*-BuOH. The dichloromethane portion (110 g) was then subjected to normal phase silica gel column chromatography and eluted with a gradient of CH₂Cl₂-MeOH (100:0 → 0:100, v/v) to give ten fractions. Fraction 2 (CH₂Cl₂-MeOH, 80:1, v/v) was subjected to further separation using ODS silica gel column chromatography eluted with 20%, 36%, 44%, 60%, and 80% MeOH in water, respectively. Subfraction 2.2 (36% methanol elution) was purified by semi-preparative HPLC (MeCN-H₂O, wavelength 280 nm) to yield compounds 1 (16.5 mg) and 2 (21.3 mg). Subfraction 2.3 (44% methanol elution) was purified by semi-preparative HPLC (MeCN-H₂O, wavelength 280 nm) to give compound 3 (10.4 mg).

The *n*-BuOH portion (280 g) was subjected to normal phase silica gel column chromatography and eluted with a gradient of CH₂Cl₂-MeOH (100:0 → 0:100, v/v) to give eight fractions. Fraction 4 was subjected to further separation using repeated ODS silica gel column and HPLC to give compounds 4 (25.0 mg) and 5 (18.6 mg).

RESULTS AND DISCUSSION

Compound 1 was obtained as a colorless needle crystal (CHCl₃), with the molecular formula C₂₂H₂₆O₈ as determined by the high-resolution electrospray ionization mass spectrometry (HRESIMS) at *m/z* 441.1518 (M + Na)⁺, indicating ten degrees of unsaturation. The ¹H-NMR spectrum of compound 1 [Table 1] displayed four aromatic proton signals at δ_H 6.59 (4H, s, H-3, H-3', H-5, H-5'), which were assigned to two 1,2,4,6-tetrasubstituted benzene rings. In addition, the ¹H-NMR spectrum also revealed four oxymethine protons at δ_H 4.74 (2H, d, *J* = 4.0 Hz, H-7, H-7') and 3.10 (2H, m, H-8, H-8'), two methylene protons at δ_H 4.28 (2H, m, H-9a, H-9'a) and 3.93 (2H, m, H-9b, H-9'b), four methoxyl groups at δ_H 3.90 (12H, s). The ¹³C-NMR spectrum displayed 22 carbon signals, including 12 aromatic

Table 1: ¹H-NMR (300 MHz) and ¹³C-NMR (100 MHz) spectral data of compounds 1 and 2 in CDCl₃

| Number | 1 | | 2 | |
|-------------------|----------------|----------------|----------------|----------------|
| | δ _H | δ _C | δ _H | δ _C |
| 1,1' | | 132.1 | | 132.0 |
| 2,2' | | 147.2 | | 147.1 |
| 3,3' | 6.59 (s) | 102.8 | 6.57 (s) | 102.8 |
| 4,4' | | 134.34 | | 134.3 |
| 5,5' | 6.59 (s) | 102.8 | 6.57 (s) | 102.7 |
| 6,6' | | 147.2 | | 147.1 |
| 7,7' | 4.74 (d, 4.0) | 86.1 | 4.78 (d, 7.0) | 86.0 |
| 8,8' | 3.10 (m) | 54.4 | 3.09 (m) | 54.3 |
| 9a, 9'a | 4.28 (m) | 71.8 | 3.89 (m) | 71.8 |
| 9b, 9'b | 3.93 (m) | | 4.26 (m) | |
| -OH | 5.52 (s) | | 5.54 (s) | |
| -OCH ₃ | 3.90 (s) | 56.4 | 3.90 (s) | 56.3 |

Coupling constants (*J*) in Hz are given in parentheses; chemical shift values are expressed in ppm. NMR: Nuclear magnetic resonance

carbons, four oxygen-substituted methine (δ_C 54.4, 54.4, 86.1, 86.1), two oxygen-substituted methylene (δ_C 71.8, 71.8), and four methoxyl carbons (δ_C 56.4). All of the above spectroscopic data were similar to those of 2,6,2',6'-tetramethoxy-4,4'-bis (1,2-*cis*-2,3-epoxy-1-hydroxypropyl) biphenyl (2) except for difference of coupling constants at H-7, 7' (*J* = 4.0 Hz). In the NOESY experiment [Figure 2] on 1, correlations between H-3/H-7 or H-5/H-7, and a strong cross-peak between H-8/H_a-9 suggested the β-configuration for H-7 and H_b-9 and an α-configuration for OH-7. The relative configuration could be proposed as 2,6,2',6'-tetramethoxy-4,4'-bis (1,2-*trans*-2,3-epoxy-1-hydroxypropyl) biphenyl.

Compounds 2-5 were identified as 2,6,2',6'-tetramethoxy-4,4'-bis (2,3-epoxy-1-hydroxypropyl) biphenyl (2), 1H-indole-3-carbaldehyde (3), luteolin (4), and quercetin (5) on the basis of its spectroscopic data.^[9-11]

CONCLUSION

One new stereoisomer of biphenyl neolignan (1) with four known compounds 2-5 was isolated from the leaves of *P. villosa* Juss. The structures of all the compounds were elucidated on the basis of spectroscopic analysis and comparison with literature data.

Compounds

2,6,2',6'-Tetramethoxy-4,4'-bis (1,2-*trans*-2,3-epoxy-1-hydroxypropyl) biphenyl (1). Colorless needle crystal. HRESIMS *m/z* 441.1518 (M + Na)⁺ (calculated for C₂₂H₂₆O₈Na, 441.1520). ¹H-NMR (CDCl₃, 300 MHz): δ_H: 3.10 (2H, m, H-8, H-8'), 4.28 (2H, m, H-9a, H-9'a), 3.90 (12H, s, OMe-2, OMe-2', OMe-6, OMe-6'), 3.93 (2H, m, H-9b, H-9'b), 4.74 (2H, d, *J* = 4.0 Hz, H-7, H-7'), 5.52 (2H, brs, OH-7, OH-7'), 6.59 (4H, s, H-3, H-3', H-5, H-5'). ¹³C-NMR δ_C: 54.4 (2C, C-8, C-8'), 56.4 (4C, OMe-2, OMe-2', OMe-6, OMe-6'), 71.8 (2C, C-9, C-9'), 86.1 (2C, C-7, C-7'), 102.8 (4C, C-3, C-3', C-5, C-5'), 132.1 (2C, C-1, C-1'), 134.3 (2C, C-4, C-4'), 147.2 (4C, C-2, C-2', C-6, C-6').

2,6,2',6'-Tetramethoxy-4,4'-bis (1,2-*cis*-2,3-epoxy-1-hydroxypropyl) biphenyl (2). Colorless needle crystal. ¹H-NMR (CDCl₃, 300 MHz): δ_H: 3.09 (2H, m, H-8, H-8'), 3.89 (2H, m, H-9a, H-9'a), 3.90 (12H, s, OMe-2, OMe-2', OMe-6, OMe-6'), 4.26 (2H, m, H-9b, H-9'b), 4.78 (2H, d, *J* = 7.0 Hz, H-7, H-7'), 5.54 (2H, brs, OH-7, OH-7') 6.57 (4H, s, H-3, H-3', H-5, H-5'). ¹³C-NMR δ_C: 54.3 (2C, C-8, C-8'), 56.3 (4C, OMe-2, OMe-2', OMe-6, OMe-6'), 71.8 (2C, C-9, C-9'), 86.0 (2C, C-7, C-7'),

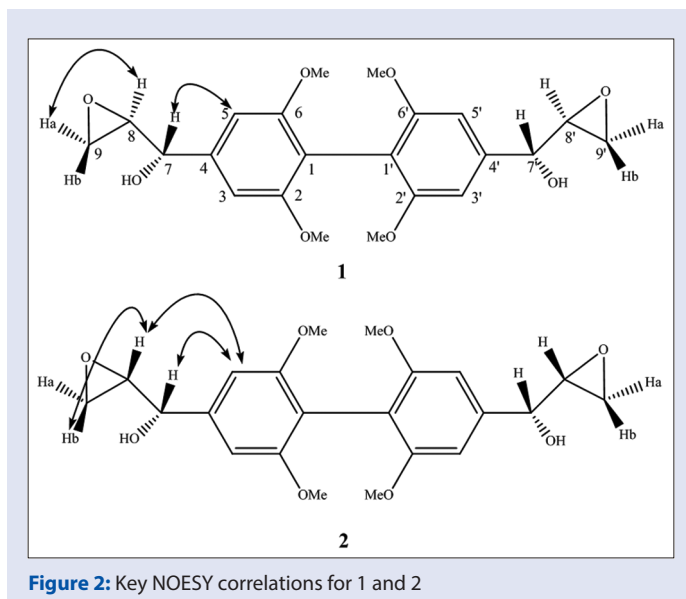


Figure 2: Key NOESY correlations for 1 and 2

102.7 (4C, C-3, C-3', C-5, C-5'), 132.0 (2C, C-1, C-1'), 134.3 (2C, C-4, C-4'), 147.1 (4C, C-2, C-2', C-6, C-6').

1H-indole-3-carbaldehyde (3). Pale yellow needle crystal. $^1\text{H-NMR}$ (dimethyl sulfoxide [DMSO], 400 MHz) δ_{H} : 7.24 (2H, m, 5,6-H), 7.52 (1H, d, $J = 8.0$ Hz, 7-H), 8.10 (1H, s, 2-H), 8.29 (1H, d, $J = 7.0$ Hz, 4-H). $^{13}\text{C-NMR}$ δ_{C} : 185.4 (C-10), 138.9 (C-8), 137.5 (C-2), 124.6 (C-9), 123.9 (C-5), 122.6 (C-6), 121.3 (C-7), 118.64 (C-3), 112.6 (C-4).

Luteolin (4). Pale yellow powder. $^1\text{H-NMR}$ (DMSO, 400 MHz) δ_{H} : 13.02 (1H, brs, 5-OH), 7.45 (1H, dd, $J = 8.5, 2.5$ Hz, 6'-H), 7.40 (1H, d, $J = 2.5$ Hz, 2'-H), 6.91 (1H, d, $J = 8.5$ Hz, 5'-H), 6.71 (1H, s, 3-H), 6.47 (1H, d, $J = 2.0$ Hz, 6-H), 6.59 (1H, d, $J = 1.6$ Hz, H-8). $^{13}\text{C-NMR}$ δ_{C} : 163.8 (C-2), 103.3 (C-3), 181.4 (C-4), 161.4 (C-5), 99.0 (C-6), 164.9 (C-7), 93.9 (C-8), 157.3 (C-9), 103.9 (C-10), 118.8 (C-1'), 113.1 (C-2'), 145.9 (C-3'), 149.9 (C-4'), 116.0 (C-5'), 121.7 (C-6').

Quercetin (5). Pale yellow powder. $^1\text{H-NMR}$ (DMSO, 400 MHz) δ_{H} : 6.20 (1H, d, $J = 2.0$ Hz, 6-H), 6.38 (1H, d, $J = 2.0$ Hz, 8-H), 6.86 (1H, d, $J = 8.4$ Hz, 5'-H), 7.56 (1H, dd, $J = 8.4, 2.0$ Hz, 6'-H), 7.65 (1H, d, $J = 2.0$ Hz, 2'-H), 9.32, 9.38, 9.62, 10.79, 12.49 (1H, s, -OH $\times 5$). $^{13}\text{C-NMR}$ δ_{C} : 146.9 (C-2), 122.1 (C-3), 175.9 (C-4), 156.3 (C-5), 98.3 (C-6), 164.0 (C-7), 93.4 (C-8), 160.9 (C-9), 103.3 (C-10), 123.3 (C-1'), 115.8 (C-2'), 148.0 (C-3'), 145.2 (C-4'), 115.2 (C-5'), 120.2 (C-6').

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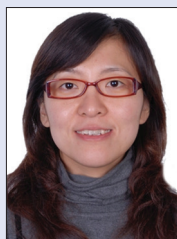
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Conflicts of interest

There are no conflicts of interest.

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