New Sesquiterpene and Polymethoxy-Flavonoids from *Artemisia annua* L

Yang Chu^{1,2}, Hongbo Wang³, Jing Chen⁴, Yue Hou⁵

¹Department of Pharmacy, the First Affiliated Hospital of China Medical University, 155 Nanjing Street, Shenyang, ²Department of the First Clinical Pharmacy, School of Pharmacy, China Medical University, 92 Bei'er Road, Shenyang, ³Department of Natural Natural Chemistry, School of Traditional Chinese Materia Medica, Shenyang Pharmaceutical University, Wenhua Road 103, Shenyang, ⁴Department of Pharmaceutics, Ningxia Medical University, Yinchuan, ⁵College of Life and Health Sciences, Department of Biochemistry and Molecular Biology, Northeastern University, Shenyang, 110819, P.R. China.

Submitted: 01-12-2013

Revised: 14-01-2014

Published: 24-07-2014

ABSTRACT

Our previous study revealed that the polymethoxy-flavonoids, as main components of *Artemisia annua*, could improve the antimalarial activity of *Artemisinin*. Here, we described the isolation, elucidation, constituent analysis, flavonoids enrichment of the extracts of *A. annua*. A total of 20 compounds were isolated including a new sesquiterpene (compound 12) and five (1, 5, 6, 7, 15) afforded for the first time from *A. annua*. The elucidation of eight flavonoids may be a useful phytochemical data and chemical foundation for further mechanism studies on improving the anti-malarial action of artemisinin. Furthermore, the antitumor activities of the compounds were assayed using four different kinds of human cancer cell lines.



Key words: Antitumor activities, artemisinin, polymethoxy-flavonoid, sesquiterpene

INTRODUCTION

Artemisia annua L. (Qinghao), a traditional Chinese herb with pharmacological functions on clearances of heat and toxic materials as well as on antimalarial bioactivity, belongs to genus *Artemisia* (Composite family) and distributes mainly in the northern parts of Guangxi and Sichuan provinces in China. However, this plant now grows wildly in Europe and America. Phytochemical studies of *A. annua* revealed the presence of terpenoids, flavonoids, aliphatic hydrocarbons, aromatic ketones, aromatic acids, phenylpropanoids,^[1] alkaloids and coumarins.^[2]

Our previous study showed that the polymethoxy-flavonoids of *A. annua* had the effects on improving the antimalarial activity of artemisinin. Using liquid chromatographic tandem mass spectroscopy method for the determination of the antimalarial drug, artemisinin, in rat plasma using arteannuin B as internal standard (I.S.), the results demonstrated that different doses of CHR significantly

Address for correspondence:

Dr. Yang Chu, Department of Pharmacy, the First Affiliated Hospital of China Medical University, 155 Nanjing Street,

and Department of the First Clinical Pharmacy, School of Pharmacy, China Medical University, 92 Bei'er Road, Shenyang 110001, China. E-mail: 15002422786@163.com increased the areas under the plasma concentration-time curve (AUC) (P < 0.01) compared with artemisinin alone and suggested that co-administration of CHR may be an efficient way to increase the anti-malarial action of artemisinin.

Here, we report the phytochemical research result of A. annua, including the isolation and elucidation of a new sesquiterpene (compound 12), together with four known sesquiterpenes (compounds 1, 2, 8 and 11), eight flavonoids (compounds 7, 13, 14, 15, 16, 17, 19 and 20), two coumarins (compounds 3 and 6) and two acetophenones (compounds 4 and 5).

MATERIALS AND METHODS

General

NMR spectra were measured with Bruker ARX-300 and ARX-600 spectrometers (Bruker Corporation, Germany), using DMSO- d_6 and CDCl₃ as solvent and TMS as an internal standard. HR-EST-MS was performed on Bruker micro TOF-Q mass spectrometer in m/z (rel.%) (Bruker Daltonics Inc., Germany). ESI-MS was performed on a Finnigan LCQ mass spectrometer (Thermo Electron, California, USA). Silica gel (200-300 mesh) and silica gel G (Qingdao Marine Chemical Group Co. Ltd, Qingdao, China) were used for column chromatography and TLC, respectively.

Extraction and Isolation

The acetone extracts of A. annua (200 g) was provided by Department of Pharmaceutics, Ningxia Medical University. Then it was subjected to HPD-100 macroporous adsorption resin and eluted with EtOH/H₂O (3:2, 7:3, 4:1, 19:1) to yield 4 fractions (F1-F4), and these fractions were subjected to silica gel column chromatography (CC), Sephadex LH-20, polyamide CC, ODS CC and PHPLC to yield 20 compounds: 3\alpha-hydroxy-1-deoxyartemisinin (1, 32.0 mg, colorless crystal in CH₂Cl₂: MeOH 3:1),^[3] α-epoxy-dihydroartemisinic acid (2, 52.0 mg, colorless crystal in CH₂Cl₂: MeOH 1:1),^[4] scopoletin (3, 116.0 mg, light yellow crystal in CH₂Cl₂: MeOH 2:1),^[5] 4, 6-dihydroxy-2-methoxy acetophenone (4, 26.0 mg, colorless crystal in CH₂Cl₂: MeOH 3:1) (Brown, 1992),^[6] 6-hydroxy-2,4-dimethoxy acetophenone (5, 17.2 mg, colorless crystal in CH₂Cl₂: MeOH 1:1) (Brown, 1992),^[6] esculetin (6, 13.6 mg, light yellow crystal in CH₂Cl₂: MeOH 2:1),^[5] quercetagetin-3,6,3',4'-tetramethyl ether (7, 24.4 mg, yellow crystal in CH₂Cl₂: MeOH 2:1),^[3] arteannuin M (8, 98.4 mg, colorless crystal in CH₂Cl₂: MeOH 1:1),^[7] salicylic acid (9, 11.0 mg, colorless crystal in MeOH), β -sitosterol (10, 10.0 mg, white crystal in CH₂Cl₂: MeOH 10:1), artemisinin (11, 109.1 mg, colorless crystal in CH₂Cl₂: MeOH 2:1),^[8] artemisilactone B (12, 42.0 mg, white crystal in MeOH), salvigenin (13, 114.2 mg, yellow crystal in CH₂Cl₂: MeOH 2:1),^[9] 5,4'-dihydroxy-3,3',7-trimethoxy flavone (14, 226.4 mg, yellow crystal in CH₂Cl₂: MeOH 2:1),^[10] artemetin (15, 158.6 mg, yellow crystal in CH₂Cl₂: MeOH 4:1),^[3] chrysosplenetin (16, 389.0 mg, yellow crystal in CH₂Cl₂: MeOH 2:1),^[3] rutin (17, 56.5 mg, yellow crystal in MeOH), daucosterol (18, 32.1mg), 5-hydroxy-3,7,3',4'-tetramethoxy flavone (19, 33.3 mg, yellow crystal in CH₂Cl₂: MeOH 4:1),^[11]

and 5-hydroxy-6,7,8,4'-tetramethoxy flavone (20, 12.0 mg, yellow crystal in CH₂Cl₂: MeOH 3:1).^[12]

Compounds 1-8, 11-16 and 19-20 were identified by comparison of their physical and spectroscopic data (EIMS, IR, ¹H and ¹³C NMR) with those reported in the literature [Figure 1].

In vitro cytotoxicity bioassay

A549, HL60, U87, DU145 cells were purchased from American Type Culture Collection (#HB-8065, ACTT, Manassas, VA, USA). The cells were cultured in RPMI-1640 medium (GIBCO, NY, USA) supplemented with 10% fetal calf serum (FCS) (Shengma Yuanheng, Beijing, China), 100 mg/L streptomycin, 100 IU/mL penicillin, and 0.03% L-glutamine, and maintained at 37°C with 5% CO₂ in a humidified atmosphere.

Compounds 1-20 were resolved in dimethyl sulfoxide (DMSO) to make a stock solution. The DMSO concentration was kept below 0.10% throughout the cell culture period and did not exert any detectable effect on cell growth or cell death. The Hela, MCF-7 and HepG2 cells were incubated at 6×10^3 cells/well in 96-well plates, respectively. The cells were incubated with the five compounds at 1, 10, 50, 100 µM for 48 hours. Cell growth was measured by a 3-(4, 5-dimethylthiazol-2-yl)-2, 5-diphenyltetrazolium bromide (MTT) assay. The percentage of cell growth inhibition was calculated as follows:

Cell growth inhibition (%) = $[A_{490}(\text{control}) - A_{490}(\text{compound})]/$ [$A_{490}(\text{control})$] ×100

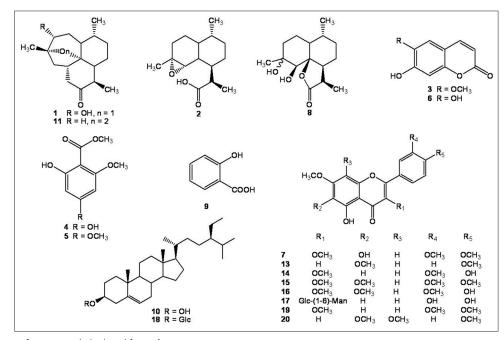


Figure 1: Structures of compounds isolated from A. annua

RESULTS AND DISCUSSION

Compound 12 was obtained as a white crystal (MeOH). The molecular formula was established as C15H22O3 on the basis of ion peaks at m/z 273.1466 [M + Na]+ (calcd. for C15H22O3Na, 273.1466) and 251.1640 [M + H]+ (calcd. for C15H23O3, 251.1646) in the HR-ESI-MS, with the help of NMR spectra. Its IR spectra showed absorption bands for hydroxyl groups (3427 cm-1), -CH2- (2921 cm-1), a terminal double bond (1630 cm-1) and a six-membered ring lactones unit (1720 cm-1). In the 1H NMR spectrum of 12 [Table 1], two methyl groups at δ 1.53 and δ 0.91 (3H, d, J = 5.4 Hz), a O-bearing methylidyne group at δ 4.47 (1H, d, J = 12.0 Hz), one pair of olefinic proton signals at δ 4.88/4.73, were observed. The 13C NMR spectrum [Table 1] of 12 showed 15 carbon signals that could be assigned to the sesquiterpene moiety. A lactone carbonyl (δ 175.1), a pair of olefinic signals with exo-cyclic double bond at δ 147.8 (101.9), which were assigned C-4 (15), three tertiary carbon signals at δ 49.0/46.2/41.4, a O-bearing tertiary carbon signal at δ 79.3 and a O-bearing quaternary carbon signal at δ 87.8, were observed in the 13C NMR spectrum of 12. Thus, these data above allowed the skeleton of 12 to be deduced as a cadinanolide-type sesquiterpene derivative. Furthermore, an examination of the 2D-NMR spectra (HMQC, HMBC) of 12 indicated that it was similar to artemisilactone,^[13] but differed from artemisilactone at C-11 and C-4 [Table 1 and Figure 2]. The HMBC experiment showed 1H/13C correlations between H-15 and C-3, C-5, between H-14 and C-1, C-9, and between H-13 and C-7, C-12. Because the H-C (1), H-C (6) and 14-Me of cadinanolide-type are defined in the α -configuration, H-C (5) is defined in the β -configuration. Additionally, the NOESY experiment showed correlations between H-1 and H-2, 6, 7, 9, 14, H-5 and H-2, 3, 8, 10, H-6 and H-1, 2, 7, 13, H-7 and H-1, 6, 8, 9, 13, H-13 and H-7, H-14 and H-1, 2, 9, 10.So 7-H and 13-Me were defined as α-configuration and 13-OH was defined as β -configuration. As a result, those elucidation allowed the structure of compound 12 [Figure 2] to be deduced.

Using A549, HL60, U87 and DU145 cell lines, the antitumor activities of compounds 1-20 were evaluated in vitro. Compounds 16 showed significant cytotoxic activities against the A549, HL60 and U87 cell lines with IC50 value at 15.76, 13.65 and 22.37 μ M, respectively. Compounds 4 and 5 exhibited moderate antitumor activities against HL60 cell line with an IC50 value at 84.07 and 50.60 μ M.

CONCLUSION

The finding of the new sesquiterpene (compound 12) will provide reference for the taxonomy of A. annua, and the

Table 1: The ¹H-NMR and ¹³C-NMR spectroscopic data and HMBC correlations of compound 12 (300 MHz for ¹H; 75 MHz for ¹³C, CDCl₃)

•		· · · ·	31
Position	δΗ	δC	HMBC
1	1.61	41.4	C-3, 5, 7, 9, 14
2α	1.98 (1H, m)	28.8	C-4, 6, 10
2β	1.47 (1H, m)		
3α	2.02 (1H, m)	29.5	C-1, 5, 15
3β	2.19 (1H, m)		C-1, 3, 7, 15
4	-	147.8	
5	4.47	79.3	
6	2.04	49.0	C-2, 4, 8, 10, 11
7	2.22	46.2	C-1, 5, 9, 12, 13
8α	1.84 (1H, m)	24.0	C-6, 10, 11
8β	1.31 (1H, m)		
9α	1.02 (1H, m)	34.2	C-1, 7, 14
9β	1.74 (1H, m)		
10	1.56	28.6	C-2, 6, 8
11	-	87.8	
12	-	175.1	
13	1.53	26.1	C-7, 12
14	0.91	19.9	C-1, 9
15a	4.73 (1H, s)	101.9	C-3. 5
15b	4.88 (1H, s)		
	The second se		

HMBC: Heteronuclear multiple bond correlation

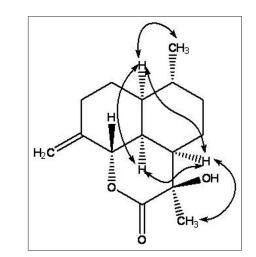


Figure 2: Key NOESY correlations of compound 12

flavonoids may be a useful phytochemical data and chemical foundation for further mechanism studies on improving the anti-malarial action of artemisinin. Furthermore, the characteristic constituent 16, polymethoxy substituted flavonoid, is a potential antitumor agent.

REFERENCES

- 1. Brown GD. The Biosynthesis of Artemisinin and the Phytochemistry of Artemisia annua L. Molecules 2010;15:7603-98.
- Bhakuni RS, Jain DC, Sharma RP, Kumar S. Secondary metabolites of Artemisia annua and their biological activity. Curr Sci 2001;80:35-48.
- 3. Chen J, Zhou YB, Zhang X, Huang L, Sun W, Wang JH. Cemical

constituents of Artemisia annua L. J Shenyang Pharm Univ 2008;25:866-70.

- Sy LK, Zhu NY, Brown GD. Syntheses of dihydroartemisinic acid and dihydro-epi-deoxyarteannuin B incoroprating a stable isotope label at the 15-position for studies into the biosynthesis of artemisinin. Tetrahedron 2001;57:8495.
- Xiao W, Li N, Ni H, Li X, Qing DG, Jia XG. Isolation and identification of chemical constituents from the whole plant of Hibiscus trionum L. J Shenyang Pharm Univ 2009;26:782-4.
- Brown GD. Two New Compounds from Artemisia annua. J Nat Prod 1992;55:1756-60.
- 7. Sy LK, Brown GD. A novel endoperoxide and related sesquiterpenes from Artemisia annua which are possibly derived from allylic hydroperoxides. Tetrahedron 1998;54:4345-56.
- Huang JJ, Nicholls KM, Chen CH, Wang Y. Two-Dimensional NMR Studies of Arteannuin. Acta Chim Sin 1987;45:305-8.
- 9. Zou JH, Yang JS. Cemical constituents of Trollius ledebouri Reicb. Chin Pharm J 2005;40:731-2.

- Yang GE, Bao L, Zhang XQ, Wang Y, Li Q, Zhang WK, et al. Studies on flavonoids and their antioxidant activities of Artemisia annua. J Chin Med Mater 2009;32:1683-6.
- Cao JQ, Dang Q, Fu HW, Yao Y, Pei YH. Isolation and identification of chemical constituents from Blumea riparia DC. J Shenyang Pharm Univ 2007;24:615-8.
- Wang QA, Wu Z, Liu L, Zou LH, Luo M. Synthesis of citrus bioactive polymethoxyflavonoids and flavonoid glucosides. Chin J Org Chem 2010;30:1682.
- 13. Zhu DY, Deng DA, Zhang SG, Xu RS. Structure of artemisilactone. Acta Chim Sin 1984;42:937-9.

Cite this article as: Chu Y, Wang H, Chen J, Hou Y. New Sesquiterpene and Polymethoxy-Flavonoids from *Artemisia annua* L. Phcog Mag 2014;10:213-6.

Source of Support: This work was supported by Liaoning Province Innovation Project of Traditional Chinese Medicine (2001-Inzyxzk-07), Conflict of Interest: None declared.